

LITERATURE CITED

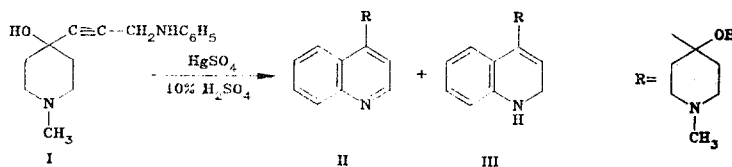
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SYNTHESIS OF 4-(1-METHYL-4-HYDROXY-4-PIPERIDYL)QUINOLINE
AND 4-(1-METHYL-4-HYDROXY-4-PIPERIDYL)-1,2-DIHYDROQUINOLINE

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We have established that 1-methyl-4-(3-anilino-1-propynyl)-4-piperidol (I) undergoes cyclization to give new, difficult-to-obtain, 4-substituted quinoline derivatives II and 1,2-dihydroquinoline III when it is heated in sulfuric acid in the presence of catalytic amounts of mercuric sulfate.



1-Methyl-4-(3-anilino-1-propynyl)-4-piperidol (I) was obtained by the reaction of 107 mmole of 1-methyl-4-piperidone and 107 mmole of N-propargylaniline in 200 ml of absolute ether in the presence of 214 mmole of KOH for 7-8 h at 0-5°C. The yield of piperidol I, with mp 113-114°C [from benzene-petroleum ether (1:1)], was 70%. IR spectrum (KBr): 1600 (benzene ring), 2170 (C≡C), 3100-3150 (OH), and 3315 cm⁻¹ (NH). PMR spectrum (CDCl₃): 2.06 (3H, s, NCH₃), 3.8 (2H, d, J = 4 Hz, CH₂NH), 4.62 (1H, broad s, OH), and 6.5-7.35 ppm (5H, m, aromatic protons).

4-(1-Methyl-4-hydroxy-4-piperidyl)quinoline (II) and 4-(1-methyl-4-hydroxy-4-piperidyl)-1,2-dihydroquinoline (III) were formed by heating 84 mmole of piperidol I in 100 ml of 10% H₂SO₄ in the presence of 1 g of HgSO₄ for 5 h at 70-80°C. The precipitate that formed after cooling and neutralization of the reaction mixture was removed by filtration and separated with a column packed with aluminum oxide (the eluent was CHCl₃ saturated with NH₃) to give 60% of quinoline II with mp 181-182°C (from benzene). IR spectrum (CHCl₃): 1580 and 1590 (quinoline ring); 3590 cm⁻¹ (OH). PMR spectrum (CDCl₃): 2.03 (3H, s, NCH₃), 5.05 (1H, s, OH), 6.87 (1H, d, J = 4 Hz, 3-H), 7.37 (2H, m, 6-H and 7-H), 7.87 (1H, d, J = 8 Hz, 5-H), 8.30 (1H, d, J = 4 Hz, 8-H), and 8.79 ppm (1H, d, J = 7 Hz, 2-H). The yield of dihydroquinoline III, with mp 120-121°C (from benzene), was 10%. IR spectrum (CHCl₃): 1685 (benzene ring), 1610 (C=C), 3460 (NH), and 3585 cm⁻¹ (OH). PMR spectrum (d₆-DMSO): 2.16 (3H, s, NCH₃), 3.32 (2H, d, J = 4 Hz, 2-H), 3.91 (1H, s, OH), 5.75 (1H, t, J = 3 Hz, 4-H), and 6.50-7.35 ppm (4H, m, 5-H, 6-H, 7-H, 8-H). The results of elementary analysis of I-III were in agreement with the calculated values.

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